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(54) PROCEDE POUR LA PREPARATION DE TRIAZOLYLISOPROPANOLS

(54) PROCESS FOR THE PREPARATION OF TRIAZOLYL ISOPROPANOLS

$$N - CH_2 - C - CH_2 - X$$

$$N = 0$$

$$V = 0$$

$$V$$

(57) Méthode améliorée d'obtention de fluconazole par réaction d'un intermédiaire halohydrinique avec un composé du type 4-amino-1,2,4-triazole de formule II (voir formule II) où X est le fluor, le chlore le brome ou l'iode, et par désamination à l'acide nitreux. Le fluconazole est un médicament antifongique efficace.

(57) An improved method for the preparation of fluconazole is described, by reacting an halohydrinic intermediate with 4-amino-1,2,4-triazole compound of formula II (see formula II) wherein X is fluorine, chlorine, bromine or iodine and subsequent deamination with nitrous acid. Fluconazole is useful as a antimycotic drug.

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# PROCESS FOR THE PREPARATION OF TRIAZOLYL ISOPROPANOLS

The present invention refers to a process for the preparation of 2-(2,4-difluorophenyl)-1,3-bis-(1H,1,2,4-triazol-1-yl)-2-propanol, of formula I

$$N - CH_2 - C - CH_2 - N$$

$$N = \frac{1}{E}$$
(I)

The compound I, also known with the name of fluconazole, is an antimycotic drug, disclosed in GB 2099818.

The known processes for the preparation of compounds I are characterized by the opening of an epoxidic intermediate of formula

with 1,2,4-triazole.

This reaction, however, is not selective, and yields the isomer 2-(2,4-difluorophenyl),1-(1H,1,2,4-triazol-1-yl),3-(4H,1,2,4-triazol-4-yl),2-propanol.

It has now been found that the compound I may be selectively obtained by reacting an halohydrin of formula II

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wherein X is fluorine, chlorine, bromine or iodine with 4-amino-1,2,4-triazole to give the compound III

$$\begin{array}{c|c}
N & CH_2 - C - CH_2 - N \\
N & F
\end{array}$$

$$\begin{array}{c|c}
N - NH_2 \\
N & N - NH_2
\end{array}$$

$$\begin{array}{c|c}
X - NH_2 \\
N & N - NH_2
\end{array}$$

$$\begin{array}{c|c}
X - NH_2 \\
N & N - NH_2
\end{array}$$

wherein X is above defined which, by reaction with nitrous acid in aqueous or alcoholic-aqueous medium, yields the compounds I with high yields and purity.

The compound III is new and it is a further object of the invention, as an intermediate.

The compound II can be easily prepared (a) from 2,4-difluorobenzene, magnesium bromide, by reaction with 1,3-dichloroacetone (Synthesis 1983,647) and then with 1H-1,2,4-triazole or (b) from α-chloro-2,4-difluoro-acetophenone by reaction with (1H-1,2,4-triazole-1) methyl magnesium chloride (Synthesis 1983,647) or (c) from 1-[2-(2,4-difluorophenyl)-2,3-epoxypropyl]-1H-1,2,4-triazole by reaction with hydrohalogen acids.

The reaction between compound II and 4-amino-1,2,4-triazole is preferably carried out in inert solvents such as  $C_1$ - $C_5$  alcohols, ketones, esters,

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ethers.

The following examples further illustrate the invention.

### EXAMPLE 1

- 5 2-(2,4-Difluorophenyl),l-(lH,1,2,4-triazol-l-yl),3-(4H,4-amino,l,2,4-triazonium-l-yl)2-propanol,bromide (III)
- 6.4 g of 2(2,4-difluorophenyl),1-bromo,3(1H,1,2,4-triazol-l-yl)-2-propanol, are refluxed in 100

  ml of isopropanol with 5.1 g of 4-amino-l,2,4-triazole
  for 8 hours. The reaction mixture is cooled to 0°C and
  the crystallized product is filtered. The crude wet
  product, so obtained, is refluxed with 50 ml of
  isopropanol, then refluxed, filtered and dried under
  vacuum at 40°C.

6.3 g (77.8%) of the title product are obtained.

#### EXAMPLE 2

## 2-(2,4-Difluorophenyl),1,3-bis-(1H,1,2,4-triazol-1-yl)-2-propanol (I)

6.3 g of the product obtained in the Example 1 are 20 dissolved in 60 ml of water and, cooling to 5°C, added with 1.8 g of concentrated hydrochloric acid. The solution is treated, at temperatures between 0 and 5°C, with a solution of 1.2 g of sodium nitrite in 6 ml of The reaction is continued at same the 25 temperature for 30 minutes and then for at least 1 hour at 20°C. The so obtained solution is added with 500 mg of active charcoal and filtered. The so obtained clear solution is treated with concentrated ammonia up to pH 9 keeping the temperature at 20°C. When the product 30 precipitation starts, the solution is cooled to 5°C for at least 5 ml of water. The obtained crude product is crystallized from 25 ml of isopropanol. The filtered product is washed with 5 ml of cold isopropanol, dried at 40°C under vacuum.

4.1 g (85.4%) of the title product, having the same elemental analysis, mass, IR and NMR spectrum as a product sample obtained according to GB 2099818.

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#### CLAIMS

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 A process for the preparation of the compound of formula I

5  $N - CH_2 - C - CH_2 - N$   $N = CH_2 - C -$ 

which comprises the reaction of a compound of formula

 $\begin{array}{c|c}
 & \text{CH} \\
 & \text{N} - \text{CH}_2 - \text{C} - \text{CH}_2 - \text{X} \\
 & \text{N} = \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 \\
 & \text{CH}_2 - \text{C} - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 \\
 & \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 \\
 & \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 \\
 & \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 \\
 & \text{CH}_2 - \text{CH}_2 -$ 

wherein X is fluorine, chlorine, bromine or iodine with 4-amino-1,2,4-triazole to give the compound of formula III

 $\begin{array}{c|c}
 & \text{CH} & \text{CH}_2 - \text{C} - \text{CH}_2 - \text{N} & \text{NH}_2 \\
 & \text{N} - \text{CH}_2 - \text{C} - \text{CH}_2 - \text{N} & \text{N} - \text{NH}_2 \\
 & \text{F} & \text{CIII}
\end{array}$ 

which is then reacted with nitrous acid.

- 2. A process according to claim 1 characterized in that a compound II wherein X-is bromine is used.
- 3. Compound of formula III

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$$N - CH_2 - C - CH_2 - N - NH_2$$

$$N = N - NH$$

wherein X is fluorine, chlorine, bromine or iodine.